This article was downloaded by:

On: 22 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713454007

Triterpenoids from the leaves of *Toona ciliata*

Jing Ning^{ab}; Hong-Ping He^a; Shi-Fei Li^{ab}; Zhao-Liang Geng^{ab}; Xin Fang^{ab}; Ying-Tong Di^a; Shun-Lin Li^a; Xiao-Jiang Hao^a

^a State Key Laboratory of Phytochemistry and Plant Resources in West China,, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming, China ^b Graduate University of the Chinese Academy of Sciences, Beijing, China

Online publication date: 15 June 2010

To cite this Article Ning, Jing , He, Hong-Ping , Li, Shi-Fei , Geng, Zhao-Liang , Fang, Xin , Di, Ying-Tong , Li, Shun-Lin and Hao, Xiao-Jiang(2010) 'Triterpenoids from the leaves of *Toona ciliata*', Journal of Asian Natural Products Research, 12:6,448-452

To link to this Article: DOI: 10.1080/10286020.2010.493329 URL: http://dx.doi.org/10.1080/10286020.2010.493329

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



ORIGINAL ARTICLE

Triterpenoids from the leaves of Toona ciliata

Jing Ning^{ab}, Hong-Ping He^a, Shi-Fei Li^{ab}, Zhao-Liang Geng^{ab}, Xin Fang^{ab}, Ying-Tong Di^a, Shun-Lin Li^a and Xiao-Jiang Hao^a*

^aState Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming, China; ^bGraduate University of the Chinese Academy of Sciences, Beijing, China

(Received 30 March 2010; final version received 12 May 2010)

One new limonoid, toonaciliatone A (1), and one new tirucallane-type triterpenoid, toonaciliatine A (4), along with three known compounds, methyl- 3β -acetoxy-1-oxomelic-14(15)-enate (2), perforin A (3), and cholest-14-ene-3,7,24,25-tetrol-21,23-epoxy-21-methoxy-4,4-trimethyl-3-(3-methyl-2-butenoate) (5), were isolated from the leaves of *Toona ciliata*. The structures of the new compounds were established by spectroscopic methods.

Keywords: Toona ciliata; limonoids; triterpenoid; tirucallane-type

1. Introduction

The plant of Toona ciliata Roem. var. ciliata (Meliaceae) is a rich source of structurally intriguing limonoids with diverse bioactivities [1-4]. Until now, a series of limonoids exhibiting different carbon frameworks and oxygenated patterns have been reported from this plant [5-8]. With the aim of searching for structurally unique and bioactive natural products, especially limonoids from the Meliaceae family, the chemical components of the leaves of T. ciliata have been investigated to give one new limonoid (1) and one new triterpenoid (4) (Figure 1). The isolation and structural elucidation of the new compounds and the cytotoxicity evaluation of all the isolated compounds are reported herein.

2. Results and discussion

Toonaciliatone A (1) was obtained as a white amorphous powder. The molecular

formula was determined to be C₂₆H₃₄O₅ by the $[M + Na]^+$ ion peak at m/z449.2303 in the HR-ESI-MS. The IR spectrum exhibited absorptions ascribable to hydroxyl (3437 cm⁻¹) and carbonyl (1724 cm⁻¹) groups. Its ¹H NMR spectrum exhibited the presence of five quaternary methyls at $\delta_{\rm H}$ 0.81 (3H, s, Me-18), 1.04 (3H, s, Me-19), 1.11 (3H, s, Me-30), 1.34 (3H, s, Me-29), and 1.41 (3H, s, Me-28), two olefinic protons at $\delta_{\rm H}$ 5.90 (1H, d, $J = 10.0 \,\mathrm{Hz}$, H-2), and 7.05 $(1H, d, J = 10.0 \,\text{Hz}, H-1)$, along with a $\beta\text{-substituted}$ furan ring. The ^{13}C NMR spectrum (Table 1) of 1 further showed the presence of two ketonic carbonyls ($\delta_{\rm C}$ 206.2 and 221.6) and two oxygenated carbons ($\delta_{\rm C}$ 67.4 and 74.7). Careful investigation of the ¹H and ¹³C NMR spectral data of 1 established that it was quite similar to those reported for 6α acetoxy-14 β ,15 β -epoxy-azadirone [5]. Comparison of the 1D NMR spectral data

^{*}Corresponding author. Email: haoxj@mail.kib.ac.cn

Figure 1. Structures of compounds 1-5.

Table 1. 13 C NMR spectral data of **1** and **4** in CDCl₃ (δ in ppm, 100 MHz).

C	1	4	C	1	4
1	157.5 (d)	33.3 (t)	20	122.6 (s)	45.3 (d)
2	126.3 (d)	22.7 (t)	21	142.9 (d)	104.3 (d)
3	206.2 (s)	77.0 (d)	22	110.7 (d)	30.8 (t)
4	45.6 (s)	36.2 (s)	23	140.2 (d)	80.3 (d)
5	44.6 (d)	41.8 (d)	24		77.9 (d)
6	67.4 (d)	23.5 (t)	25		144.6 (s)
7	74.7 (d)	72.1 (d)	26		112.6 (t)
8	42.1 (s)	44.4 (s)	27		18.4 (q)
9	44.6 (d)	41.6 (d)	28	20.2 (q)	21.7 (q)
10	40.3 (s)	37.6 (s)	29	32.1 (q)	27.8 (q)
11	18.1 (t)	16.3 (t)	30	18.4 (q)	27.7 (q)
12	34.1 (t)	32.8 (t)	1'		166.5 (s)
13	42.1 (s)	46.6 (s)	2'		116.9 (d)
14	62.2 (d)	162.1 (s)	3′		155.7 (s)
15	221.6 (s)	119.6 (d)	4'		27.4 (q)
16	43.3 (t)	34.9 (t)	5′		20.2 (q)
17	38.1 (d)	52.5 (d)	OMe		54.7 (q)
18	27.7 (q)	19.8 (q)			
19	20.9 (q)	15.2 (q)			

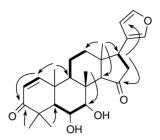


Figure 2. ${}^{1}H-{}^{1}H$ COSY (\longrightarrow) and selected HMBC (\rightarrow) correlations of 1.

with those of 6α -14 β ,15 β -epoxy-azadirone, and the analysis of the HSOC, HMBC, and ¹H-¹H COSY data indicated that the two compounds had the same carbon backbone. The differences between them were that the 14\beta,15\beta-epoxy ring and 6,7-diacetyl groups in the latter were replaced by 15-ketone and 6,7-dihydroxy groups in compound 1, respectively. This conclusion was also supported by the HMBC correlations of H-14 and H-16 with C-15 ($\delta_{\rm C}$ 221.6, s) and the cross-peaks of OH-7/H-7 and H-7/H-6 in the ${}^{1}H-{}^{1}H$ COSY spectrum (Figure 2). In the ROESY experiment, correlations of H-14/OH-7 and Me-30/H-6 indicated that OH-6 and OH-7 were of α -orientation. Thus, the structure of 1 was completely elucidated.

Triterpenoid (4) was obtained as a colorless oil. Its molecular formula was determined as C₃₆H₅₆O₆ by HR-ESI-MS data at m/z 607.3969 [M + Na]⁺. Its IR absorption bands at 3442 and 1656 cm⁻¹ suggested the presence of hydroxyl and double bond functions, respectively. In the ¹H NMR spectrum, the signals of eight methyls and one β-substituted furan ring were observed, together with two olefinic protons at $\delta_{\rm H}$ 5.46 (1H, d, $J=9.5\,{\rm Hz}$, H-15) and 5.77 (1H, s, H-2'). Its ¹³C NMR spectrum further showed the presence of four oxygenated methines ($\delta_{\rm C}$ 77.0, 80.3, 77.9 and 104.3) and one ester carbonyl ($\delta_{\rm C}$ 166.5). Comparison of the NMR spectral data of 4 with those of 5 found an overall similarity, except for the apparent different chemical shifts of

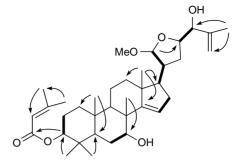


Figure 3. Selected ${}^{1}H-{}^{1}H$ COSY (\longrightarrow) and HMBC (\rightarrow) correlations of 4.

C-25, C-26, and C-27 [9]. The difference between these two compounds was due to an occurrence of a double bond [$\delta_{\rm H}$ 5.05 (1H, s) and 4.93 (1H, s); $\delta_{\rm C}$ 144.6 (s) and 112.6 (t)] between C-25 and C-26 in 4. Extensive 2D NMR experiments (HSQC, HMBC, $^{1}\text{H}-^{1}\text{H}$ COSY, and ROESY), especially the HMBC correlations of Me-27 with C-24, C-25, and C-26, confirmed the structure of 4 (Figure 3).

The three known compounds were identified as methyl- 3β -acetoxy-1-oxomelic-14(15)-enate (2) [10], perforin A (3) [11], and cholest-14-ene-3,7,24,25-tetrol-21,23-epoxy-21-methoxy-4,4,8-trimethyl-3-(3-methyl-2-butenoate) (5) by comparison of their 1D NMR data with those in the literature [9].

Compounds 1–5 were tested for *in vitro* inhibitory activities against HL-60, SMMC-7721, A549, SK-BR-3, and PANC-1 human tumor cell lines (for more details, see Supporting Information). The results indicated that all the compounds were inactive against the above tumor cell lines (with $IC_{50} > 40 \mu M$).

3. Experimental

3.1 General experimental procedures

IR spectra were recorded on a Bio-Rad FTS-135 spectrometer with a KBr disk. Optical rotations were measured with a Perkin-Elmer model 241 polarimeter. NMR spectra were measured on either a Bruker AM-400

or a DRX-500 instrument. ESI-MS and HR-ESI-MS spectra were measured with a Finnigan MAT 90 instrument and a VG Auto Spec-3000 spectrometer, respectively. Column chromatography was performed on silica gel (90–150 µm; Qingdao Marine Chemical Inc., Qingdao, China), MCI gel (CHP20P, 75-150 µm; Mitsubishi Chemical Industries Ltd, Japan), Sephadex LH-20 (40–70 µm; Amersham Pharmacia Biotech AB, Uppsala, Sweden), and C18 reversephase silica gel (150-200 mesh; Merck, Darmstadt, Germany). Semi-preparative HPLC was performed on a Zorbax SB-C-18 column (i.d. 9.4×250 mm; Agilent Co. Ltd, Santa Clara, USA). TLC plates were pre-coated with silica gel GF-254 and HF-254 (Qingdao Haiyang Chemical Plant, Qingdao, China).

3.2 Plant material

The leaves of *T. ciliata* were collected in Wenshan, Yunnan Province, China, in July 2007, and were identified by Prof. Heng Li. A voucher specimen (No. 2007-5-10) has been deposited in the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany.

3.3 Extraction and isolation

The air-dried and powdered leaves (2.6 kg) were extracted with EtOH (95%) three times. The extracts were then suspended in H₂O and further extracted with petroleum ether (PE), EtOAc, and n-BuOH, respectively. The EtOAc extracts (80g) were subjected to silica gel CC eluting with PE-acetone (1:0, 9:1, 7:3, 0:1) to afford four fractions (A1-A4). Fraction A3 (9.2 g) was applied to MCI material $(MeOH-H_2O, 70/30 \text{ to } 100/0) \text{ to give}$ four fractions, B1-B4. Fraction B3 (820 mg) was first subjected to Sephadex LH-20 (MeOH) and then to silica gel CC eluting with CHCl₃-acetone (100:1, 300 ml) and yielded 3 (24 mg). Fraction B2 (1.2 g) was subjected to Sephadex LH-20 (MeOH) to afford three fractions (C1–C3). Fraction C1 (56 mg) afforded 1 (20 mg) by semi-preparative HPLC (CH₃OH–H₂O, 60–40) (flow rate, 3.0 ml/min, detection, UV 210, 254 nm). Fraction B4 (1.6 g) was first subjected to reverse-phase C-18 silica gel (CH₃OH–H₂O, 70/30 (4 liters), 80/20 (2 liters), 90/10 (2 liters)) to afford three fractions (D1–D3). Fraction D2 (980 mg) was subjected to silica gel (CHCl₃–acetone: 300:1 (120 ml), 200:1 (100 ml), 80:1 (400 ml)) to afford compounds 2 (10 mg), 4 (10 mg), and 5 (50 mg), respectively.

3.3.1 Toonaciliatone A (1)

A white amorphous powder. $[\alpha]_D^{25} + 22.2$ $(c = 0.09, \text{CHCl}_3)$. IR (KBr) ν_{max} : 3437, 1724, 1668, 756 cm⁻¹. ESI-MS *m/z*: 449 $[M + Na]^+$. HR-ESI-MS m/z: 449.2312 $[M + Na]^+$ (calcd for $C_{26}H_{34}O_5Na$, 449.2303). ¹H NMR (CDCl₃, 400 MHz) $\delta_{\rm H}$: 0.81 (3H, s, Me-18), 1.04 (3H, s, Me-19), 1.11 (3H, s, Me-30), 1.34 (3H, s, Me-29), 1.41 (3H, s, Me-28), 1.31 (1H, dd, J = 13.6, 6.0 Hz, H_B-12), 1.38 (1H, dd, J = 11.2, 3.1 Hz, H-9), 2.04 (1H, dt, $J = 13.6, 3.1 \text{ Hz}, H_{\alpha}-12), 1.66-1.72 (2H,$ m, H-11), 2.19 (1H, d, J = 12.0 Hz, H-5), 2.54 (2H, d, J = 10.0 Hz, H-16), 2.72 (1H, s, H-14), 3.15 (1H, d, J = 4.4 Hz, OH-7), 3.50 (1H, t, J = 10.0 Hz, H-17), 3.81 (1H, t)s, H-7), 4.16 (1H, t, $J = 12.0 \,\text{Hz}$, H-6), 5.90 (1H, d, J = 10.0 Hz, H-2), 7.05 (1H, d, $J = 10.0 \,\text{Hz}, \,\text{H}-1$), 6.30 (1H, s, H-22), 7.29 (1H, s, H-23), 7.41 (1H, s, H-21). ¹³C NMR data (CDCl₃, 100 MHz): see Table 1.

3.3.2 Toonaciliatine A (4)

A colorless oil. $[\alpha]_{\rm D}^{25} - 151.7$ (c = 0.1, CHCl₃). IR (KBr) $\nu_{\rm max}$: 3442, 3425, 1656, 1032 cm⁻¹. ESI-MS m/z: 585 [M + H]⁺. HR-ESI-MS m/z: 607.3969 [M + Na]⁺ (calcd for C₃₆H₅₆O₆Na, 607.3974). ¹H NMR (CDCl₃, 500 MHz) $\delta_{\rm H}$: 0.85 (3H, s, Me-28), 0.91 (3H, s, Me-29), 0.91 (3H, s,

Me-19), 1.03 (3H, s, Me-18), 1.06 (3H, s, Me-30), 1.77 (3H, s, Me-27), 1.89 (3H, s, Me-4'), 2.17 (3H, s, Me-5'), 1.26-1.32 $(1H, m, H_{\alpha}-1), 1.35-1.40 (1H, m, H_{\beta}-1),$ 1.50-1.54 (1H, m, H_{β} -12), 1.51-1.55 $(1H, m, H_{\beta}-11), 1.61-1.65$ (1H, m, H_{α} -12), 1.62–1.66 (1H, m, H_{α} -2), 1.70– 1.74 (1H, m, H_{α} -11), 1.73–1.77 (1H, m, H_{B} -22), 1.74–1.78 (2H, m, H_{2} -16), 1.88– 1.93 (1H, m, H_{α} -22), 1.99 (1H, d, $J = 3.1 \,\text{Hz}, \, \text{H-}17$), 2.02–2.06 (1H, overlapped, H-9), 2.02-2.06 (1H, overlapped, H-5), 2.14–2.20 (2H, m, H-16), 2.22–2.27 (1H, m, H-20), 3.36 (3H, s, 21-OMe), 3.83 (1H, d, $J = 4.5 \,\text{Hz}$, H-24), 3.91 (1H, s, H-7), 4.20–4.25 (1H, m, H-23), 4.69 (1H, s, H-3), 4.76 (1H, d, J = 4.1 Hz, H-21), 4.93 (1H, s, H_a-26), 5.05 (1H, s, H_b-26), 5.46 (1H, d, J = 9.5 Hz, H-15), 5.77 (1H, s, H-2'). ¹³C NMR data: see Table 1.

Acknowledgements

This work was financially supported by grants from the Ministry of Science and Technology (2009CB940900 and 2009CB522303). We thank Prof. Heng Li for the identification of the plant material.

References

- T.R. Govindachari, G. Suresh, G. Gopalakrishnan, S. Masilamani, and B. Banumathi, *Fitoterapia* 71, 317 (2000).
- [2] W. Kraus and W. Grimminger, *Nouv. J. Chim.* **4**, 651 (1980).
- [3] W. Kraus and W. Grimminger, *Liebigs Ann. Chem.* **10**, 1838 (1981).
- [4] W. Kraus, W. Grimminger, and G. Sawitzki, *Angew. Chem.* **90**, 476 (1978).
- [5] J.O. Neto, S.M.M. Agostinho, M.F.D.G.F.D. Silva, P.C. Vieira, J.B. Fernandes, A.L. Pinheiro, and E.F. Vilela, *Phytochemistry* 38, 397 (1995).
- [6] J.O. Neto, M.F.D.G.F.D. Silva, E.R. Fo, J.B. Fernandes, P.C. Vieira, and A.L. Pinheiro, *Phytochemistry* 49, 1369 (1998).
- [7] H.D. Chen, S.P. Yang, Y. Wu, L. Dong, and J.M. Yue, J. Nat. Prod. 72, 685 (2009).
- [8] S.G. Liao, S.P. Yang, T. Yuan, C.R. Zhang, H.D. Chen, Y. Wu, Y.K. Xu, and J.M. Yue, J. Nat. Prod. 70, 1268 (2007).
- [9] K. Mitsui, H. Saito, R. Yamamura, H. Fukaya, Y. Hitotsuyanagi, and K. Takeya, *Chem. Pharm. Bull.* 55, 1442 (2007).
- [10] D.A.H. Taylor, J. Chem. Soc. (C) 18, 2439 (1969).
- [11] K. Kamiuchi, K. Mitsunaga, K. Koike, Y. Ouyang, T. Ohmoto, and T. Nikaido, *Heterocycles* 43, 653 (1996).